

The formation of iron oxide – copper – gold (IOCG) and iron oxide – apatite (IOA) deposits: Insights from the chemistry of actinolite, apatite, magnetite and pyrite

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Source: Reich et al., 2022





Source: Reich et al., 2022

Element	Deposit type	Grade (median)	Crustal abundance	Clarke value
Cu	IOCG	0.5 %	0.027 %	~200
Fe	IOA	>40 - 50%	5 %	>8 - 10







Geological resources, Millions of tonnes









Punta del Cobre district





Candelaria IOCG deposit





Punta del Cobre district



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Source: del Real

Candelaria IOCG deposit





...transitions to chalcopyrite-magnetite-biotite –K-feldspar ± pyrite ± actinolite mineralization/alteration dated at ca. 115 Ma that represents the main Cu mineralization stage in the area.

Pervasive magnetite-actinolite (sulfide-free) alteration assemblages at deeper levels along with regional-scale Na ± Ca-rich hydrothermal alteration



Candelaria metal grades







Source: del Real et al. (2023)

Magnetite, Candelaria



Source: Rodriguez-Mustafa

Magnetite, El Laco







Magnetite, Los Colorados





Magnetite, Los Colorados



Magnetite chemistry and temperature



These data highlighted by the grey arrow are from a single dill core at El Laco.

These data show a systematic trend of decreasing trace element concentrations from depth to surface.



Magnetite chemistry and temperature



Source: Palma et al., 2020

Range of temperature estimated from mineral equilibria, stable isotope and fluid inclusion studies (~300–1,000 $^{\circ}\text{C})$



Magnetite chemistry and temperature





Apatite chemistry



Source: La Cruz et al., 2020

Apatite chemistry



 87 Sr/ 86 Sr and ε_{Nd} of F-rich apatite indicate a magmatic source for the ore fluid that precipitated fluorapatite. (Palma, 2019).

El Laco apatite up to 1 wt% S, indicating precipitation from oxidized fluid. (La Cruz, 2020)



Actinolite chemistry





Actinolite chemistry





Actinolite chemistry



Source: del Real

Candelaria metal grades and temperatures



** ± 20°C



Magnetite Fe and O stable isotopes





Source: Childress et al., 2020

Magnetite Fe and O stable isotopes



Source: Rodriguez-Mustafa et al., 2022

Pyrite chemistry



Source: del Real et al.

Pyrite chemistry



Source: Reich et al., 2016

Pyrite chemistry





Source: Reich et al., 2016

Mineral chemistry and textures are consistent with a hydrothermal origin and record multiple pulses of magmatic-hydrothermal fluid.



(A)



Extension in a back-arc environment allows intermediate to mafic magmas to ascend to shallow levels of the crust.

The high temperature, low viscosity magmas allow exsolved magmatic-hydrothermal fluid to passively leave the magma body.

DE EXPLORAÇÃO MINERA





Early hydrothermal fluids evolve from intermediate to mafic magmas and would be enriched in dissolved Fe as well as Ca, Mg and Si to form magnetite + actinolite during cooling and interaction with the host rocks.

The hypogene ore fluid precipitates magnetite and actinolite at temperatures > 650 °C.



(C)



Later, lower temperature ore fluids evolve precipitate Cu-Fe-sulfides such.

This later fluid also precipitates magnetite, actinolite, pyrite and other minor phases.

 $\mathbf{CuCl}_{(aq)} + \mathbf{H}_{2}\mathbf{O}_{(aq)} + \mathbf{Fe}_{3}\mathbf{O}_{4(mt)} = \mathbf{CuFeS}_{2(cp)} + \mathbf{H}_{2}\mathbf{O}_{(aq)}$

 $Fe_{3}O_{4(mt)} + H_{2}S_{(aq)} + O_{2(aq)} = FeS_{2(py)} + H_{2}O_{(aq)}$





AI SIMPUSIU BRASILEINU DE EXPLORAÇÃO MINERAL

ON MINERAL EXPLORATION

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Obrigado!

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